## LIST OF SYMBOLS

- A pre-exponential factor  $(s^{-1})$
- A<sub>1</sub> Percent of ash present in coal
- A<sub>2</sub> Percent of SiO<sub>2</sub> present in coal-ash
- A<sub>3</sub> Percent of MnO present in coal-ash
- $A_4$  Percent of  $P_2O_5$  present in coal-ash
- A<sub>5</sub> Percent of CaO present in coal-ash
- A<sub>6</sub> Percent of Al<sub>2</sub>O<sub>3</sub> present in coal-ash
- A<sub>7</sub> Percent of MgO present in coal-ash
- A<sub>8</sub> Percent of sulphur present in coal-ash
- A<sub>9</sub> Percent of Fe<sub>2</sub>O<sub>3</sub> present in coal-ash
- C<sub>1</sub> Percent of coal present in composite pellets
- E activation energy  $(kJmol^{-1})$
- F fraction of reduction
- F<sub>1</sub> Percent of iron ore present in composite pellets
- F<sub>2</sub> Percent of Fe<sub>2</sub>O<sub>3</sub> present in iron ore
- F<sub>3</sub> Percent of total Fe present in Fe<sub>2</sub>O<sub>3</sub>
- F<sub>4</sub> Percent of Fe loss in slag w.r.t. total Fe-input
- F<sub>5</sub> Percent of SiO<sub>2</sub> present in iron ore of composite pellets
- F<sub>6</sub> Percent of MnO present in iron ore of composite pellets
- F<sub>7</sub> Percent of P<sub>2</sub>O<sub>5</sub> present in iron ore of composite pellets
- F<sub>8</sub> Percent of Al<sub>2</sub>O<sub>3</sub> present in iron ore of composite pellets
- F<sub>9</sub> Percent of fixed carbon present in coal
- F<sub>10</sub> Percent of total iron present in iron ore
- F<sub>11</sub> Percent of other oxides present in iron ore
- F<sub>12</sub> Percent of dextrose present in composite pellets
- F<sub>13</sub> Percent of volatile materials present in coal
- F<sub>14</sub> Percent of moisture present in coal

f frequency, Hz

- $f_{\text{coal}}$  Fraction of coal present in composite pellet
- $f_0$  Fraction of oxygen present in Fe<sub>2</sub>O<sub>3</sub>
- $f_{\rm ore}$  Fraction of ore present in composite pellet
- $\rho_{\rm ore}$  Purity of iron oxide (Fe<sub>2</sub>O<sub>3</sub>)
- $f_{\rm p}$  factor of proportionality
- $f_R$  fraction of reaction
- $f_{\rm vm}$  Fraction of volatile matters present in coal
- $f_{\rm vr}$  Fraction of volatiles released during reduction at a particular temperature
- $f_{wl}$  Fractional weight loss = (weight loss of sample / initial weight of sample)
- I<sub>1</sub> Percent of Fe present in cast iron charged
- I<sub>2</sub> Percent of Fe present in cast iron produced
- I<sub>3</sub> Percent of Si present in cast iron charged
- I4 Percent of Si present in cast iron produced
- I<sub>5</sub> Percent of Mn present in cast iron charged

- $I_6$ Percent of Mn present in cast iron produced
- Percent of P present in cast iron charged  $I_7$
- Percent of P present in cast iron produced  $I_8$
- Percent of C present in cast iron charged I9
- $I_{10} \\$ Percent of C present in cast iron produced
- Percent of S present in cast iron charged  $I_{11}$
- Percent of S present in cast iron produced I12
- Percent of Cr present in cast iron scrap charged I<sub>13</sub>
- Percent of Cr present in cast iron produced I<sub>14</sub>
- Percent of Cu present in cast iron scrap charged I15
- Percent of Cu present in cast iron produced  $I_{16}$
- Percent of lime present in composite pellets  $L_1$
- Percent of SiO<sub>2</sub> present in lime  $L_2$
- Percent of CaO present in lime  $L_3$
- Percent of Al<sub>2</sub>O<sub>3</sub> present in lime L4
- Percent of MgO present in lime Ls
- rate constant (s<sup>-1</sup>) k
- K Kelvin
- $O_1$ Weight of oxygen required for oxidation of iron, kg
- Weight of oxygen required for oxidation of silicon, kg  $O_2$
- Weight of oxygen required for oxidation of manganese, kg  $O_3$
- Weight of oxygen required for oxidation of phosphorous, kg  $O_4$
- Weight of oxygen required for formation of CO gas, kg 05
- Weight of oxygen required for formation of CO2 gas, kg  $O_6$
- Weight of oxygen coming from iron ore present in composite pellets, kg  $O_7$
- Weight of oxygen require from atmosphere for oxidation of elements, kg **O**<sub>8</sub>
- Weight of oxygen required for oxidation of Cr, kg O<sub>9</sub>
- Weight of oxygen required for oxidation of Cu, kg O<sub>10</sub>
- gas constant (kJmol<sup>-1</sup>K<sup>-1</sup>) R
- Т temperature in Kelvin scale
- tpa tonnes per annum
- tpd tonnes per day
- tonnes per hour tph
- VM volatile matters
- W Weight of Fe<sub>2</sub>O<sub>3</sub> in iron ore, kg
- Weym amount of volatiles released during pyrolysis of coal
- Wi initial weight of a single pellet (g)
- Wf weight of pellet after reduction
- amount of volatiles released during TG-DTA of composite pellet W<sub>pvm</sub>
- $\Delta \hat{W}_{o}$  $\Delta W_{o}^{H}$ weight of oxygen removed from iron oxide
- weight of oxygen loss during hydrogen reduction
- W<sub>o</sub><sup>i</sup> Total weight of removable oxygen present in iron oxide
- W°. rate of oxygen loss, g/s
- weight loss during reduction (g) W<sub>r</sub>

- $W_t$  weight of pellet at time t during reduction (g)
- W1 Weight of cast iron charged, kg
- W<sub>2</sub> Weight of composite pellets charged, kg
- W<sub>3</sub> Weight of cast iron produced, kg
- W<sub>4</sub> Weight of Fe goes to slag, kg
- W<sub>5</sub> Weight of FeO goes to slag, kg
- W<sub>6</sub> Weight of SiO<sub>2</sub> goes to slag, kg
- W<sub>7</sub> Weight of MnO goes to slag, kg
- W<sub>8</sub> Weight of P<sub>2</sub>O<sub>5</sub> goes to slag, kg
- W<sub>9</sub> Weight of CaO goes to slag, kg
- W<sub>10</sub> Weight of Al<sub>2</sub>O<sub>3</sub> goes to slag, kg
- W<sub>11</sub> Weight of MgO goes to slag, kg
- W<sub>C</sub> Weight of carbon converted into CO and CO<sub>2</sub> gases, kg
- W12 Weight of CO gas formed after smelting reduction of composite pellets, kg
- W13 Weight of CO2 gas formed after smelting reduction of composite pellets, kg
- W<sub>14</sub> Weight of sulphur goes to slag, kg
- W<sub>15</sub> Weight of CaS goes to slag, kg
- W<sub>16</sub> Weight of CaO converted into CaS that goes to slag, kg
- W<sub>17</sub> Weight of other oxides except SiO<sub>2</sub>, MnO, FeO, Al<sub>2</sub>O<sub>3</sub> and P<sub>2</sub>O<sub>5</sub> from iron ore that goes to slag, kg
- W<sub>18</sub> Weight of FeO that goes to slag due to Fe<sub>2</sub>O<sub>3</sub> present in coal ash, kg
- W<sub>19</sub> Weight of dextrose present in composite pellets, kg
- W20 Weight of volatile matters present in coal of composite pellets, kg
- W<sub>21</sub> Weight of moisture present in coal of composite pellets, kg
- W<sub>22</sub> Weight of slag produced, kg
- W<sub>23</sub> Weight of Cr<sub>2</sub>O<sub>3</sub> goes to slag, kg
- W24 Weight of Cu2O goes to slag, kg
- W<sub>0</sub> Weight of iron ore present in composite pellets, kg
- W<sub>car</sub> Weight of carbon stoichiometrically required for reduction of W<sub>0</sub> kg of iron ore, kg
- W<sub>coal</sub> Weight of coal required for reduction of W<sub>0</sub> kg of iron ore present in composite pellets, kg
- $Y_1$  Weight of SiO<sub>2</sub> forms due to oxidation of Si, kg
- Y<sub>2</sub> Weight of MnO forms due to oxidation of Mn, kg
- $Y_3$  Weight of  $P_2O_5$  forms due to oxidation of P, kg
- Z atomic number
- $\Sigma$  foaming index of slag
- ρ density of slag
- $\rho_a$  apparent density of pellet
- $\rho_t$  true density of pellet
- $\alpha$  degree of reduction (pct)
- μ viscosity of slag
- γ surface tension of slag
- $\eta_{pc}$  heat transfer efficiency (pct)