BAND GAP ENGINEERING OF FUNCTIONALIZED SEMICONDUCTORS FOR PHOTOCATALYTIC

APPLICATIONS



Synopsis

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Abstract

In this thesis, we have studied the electronic and optical properties of functionalized $g - C_3 N_4$ along with possible photocatalytic reaction mechanisms mainly for hydrogen evolution reaction (HER) and oxygen evolution reaction (OER) using Density Functional Theory. Starting from the study of monolayer $g - C_3 N_4$ and its quantum confinement from bulk-phase, functionalization of $g - C_3 N_4$ by doping of non-metals (B, O, P)/loading of metals (Co, Fe, Ag) along with suitable doping-loading (B-Co, O-Fe, P-Ag) have been studied. A system with suitable band edges was screened to perform overall water splitting. Next, functionalization of $g - C_3 N_4$ by the formation of metal-semiconductor heterojunction using Cobalt-boride (Co_2B_2) in 0D and 2D form has been considered in order to understand the role of dimensionality in charge transfer across the interface. Further, the semiconductor-semiconductor heterostructure comprising of MXene (Hf_2CO_2) over $g - C_3N_4$ provides insights into band bending in OER/HER. Structural modification in the pristine $g - C_3 N_4$ by different stacking of bilayer and Li intercalation in the most favorable bilayer have also been investigated. An elaborative study of charge transfer over conventional and simultaneous water-splitting reactions using climbing image nudge elastic band (CI-NEB) shows it as a potential photocatalyst. This thesis also unravels the other photocatalytic reaction mechanisms like carbon dioxide reduction reaction and nitrogen reduction reaction over the single atom catalyst $(Co - g - C_3N_4)$ with possible reaction intermediates and reaction pathways to determine the most feasible reaction. Overall, this thesis describes the role of different functionalization over photocatalytic activities and defines strategies to engineer band gap of $g - C_3 N_4$, in order to tailor the photocatalytic performance from the variation at the atomistic level.

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Objectives

The main aim was to investigate the monolayer, functionalized monolayer, modified bilayer of $g - C_3N_4$, heterostructure of $g - C_3N_4$ with other 2D materials like MXene and MBene (2D/0D) on $g - C_3N_4$ as co-catalyst with the vdW energy correction using *ab initio* calculations and to search for new more efficient photocatalytic materials.

- 1. Influence of transition metal decoration, non-metal element doping, and loading on the electronic and optical properties have been checked.
- 2. Different types of heterostructures, mechanisms of Z-scheme, interfacial and Schottky barrier potential by charge transfer, cohesive energy, and work function in heterostructures have also been studied for a detailed understanding of the photocatalytic process.
- 3. Investigated the possible structure modification and changes in electronic properties of functionalized heterostructure by determination of ground state structure. Electronic band structure of states at different conditions with van der Waals interaction.
- 4. Photocatalytic behavior of the designed materials have been studied by calculating band gap, valence band, and conduction band edges for HER and OER, and analysis of band structure curves for the effective mass calculations to compute the rate of recombination of electrons and holes.

Thesis Outline

This thesis has been divided into seven chapters, starting with the introduction consisting of the basic functionality of photocatalysts and photocatalysis. Work done on the proposed objectives of various functionalization methods and different reaction mechanisms has been reported in Chapter 1. Chapter 2 consists of the computational methodology utilized and the reaction mechanism performed to study the photocatalytic activity of the proposed functionalized semiconductors. Electronic properties, optical properties and photocatalytic activity for metal/non-Metal loaded/doped $g - C_3N_4$ monolayer has been discussed in Chapter 3. $g - C_3N_4$ based heterostructures consists of (0D/2D)-2D metal-semiconductor junction along with 2D-2D semiconductor-semiconductor junction have been investigated and reported in Chapter 4 based on the charge transfer scheme. Chapter 5 shows a study of modified bilayer of $g - C_3N_4$ with different stacking configurations and Li-atom intercalation within the bilayer. Various pathways for carbon dioxide reduction reaction and nitrogen reduction reaction have been explored on the Co-loaded $g - C_3N_4$, a Single atom catalyst, in Chapter 6. Conclusions and future aspects of the work done over the photocatalytic applications on the functionalized $g - C_3N_4$ are discussed in the last Chapter.

1 Suitable Semiconductor Photocatalyst: An Introduction

Energy is the basic requirement of every being that ever existed on this land. From energy harnessing to its conversion, there exist various methods in the natural cycle, Photosynthesis being one of them. [1] And, humans since the discovery of fire have been looking for ways to do the same, although energy storage has emerged as the new way of conserving energy since the industrial revolution for tackling any futuristic energy crisis. Today, due to a shortage of non-renewable energy sources, time demands innovative technological advancement for enhanced solar energy conversion along with efficient materials for sustainable utilization of natural sunlight for conversion of greenhouse gases into fuels. [2] Photocatalyst, a material utilized for the rapid conversion of naturally abundant molecules into fuels is the solution to the grave energy crisis. [3]

Polymeric and polymorphic semiconductor, graphitic Carbon Nitride $(g - C_3N_4)$ has been a priority of investigation in scientific communities owing to its large surface area for reactant adsorption, rapid photogenerated charge carrier mobility, high optical absorbance in the visible spectrum, non-toxic, eco-friendly, lamellar structure with ease in chemical exfoliation, the feasibility of reactions over the surface due to presence of π -conjugation and double π -conjugation between C-N atom in the web, and two layers separated by weak van der Waals interaction, respectively. [4] Wide band gap capable of culminating intermediate states formed on the doping/loading/decoration of non-metal, metal, metallic cluster-based impurities, stable structural corrugation on reducing dimensionality or on introducing heavy atom in $g - C_3N_4$ mesh, the affinity of sustaining its electronic property on high lattice mismatch, suitable workfunction for interface charge transfer on the formation of the heterostructure, and band edges value corresponding to reduction-oxidation potential promotes capability of $g - C_3N_4$ as promising photocatalyst. [5–7]

Although, the fast recombination rate of charge carriers, localization of electrons over the C-N bond over the surface, high prohibitive overpotential for oxygen evolution reaction (OER), and band gap tending towards ultraviolet region on reducing dimensionality due to quantum confinement challenge its suitability as an efficient photocatalyst. Whereas transition metal (TM) oxides since their synthesis have shown promising applications in electrochemical photolysis, the presence of heavy metals, and transition metal sulfides despite their high optical absorbance contain the toxic sulfur which, if utilized for contaminant degradation or water splitting could affect human health. Substantial reduction in the concentration of heavy metals on the other hand will fulfill the importance of the metal oxide/sulfide-based photocatalyst while reducing the health hazard created. [8] TM and noble metals (Ag, Au, Pd, Pt) due to $p_{\pi} - d_{\pi}$ interaction between N atom of $g - C_3N_4$ and metal atom have shown the formation of charge transfer channel for adsorption site activation along with red shifting of optical absorbance for higher photogeneration of electron-hole pair. Surface plasmon resonance on light-matter interaction has been shown to increase the efficiency of photocatalysts on noble metal loading over $g - C_3N_4$. Non-metal doping in place of C/N or self-doping due to its stability increase either the concentration of electrons or holes, leading to the formation of either n-type or p-type semiconductor enhancing the mobility of electron/hole due to variation in the band structure curvature of impurity states. The synergistic effect of non-metal doping/metal loading affects the electronic, and optical properties of the pristine material while the activation of sites reduces the energy barrier between the intermediate of the reactions due to the formation of photoanodic/photocathodic sites. [9–11]

Metal-Semiconductor/Semiconductor-semiconductor heterojunction [12-14] owing to their difference in work function and band edges position governs the charge transfer directionality and specific mechanism for enhancement of charge separation capability. Charge transfer across the Schottky junction leads to balanced Fermi generating the interfacial electric field for unidirectional electron transfer as seen for various metal-semiconductor interfaces. The band edge governing the straddling, staggering, or breaking of band alignment in semiconductor-semiconductor junction leads to band bending for feasible migration of electron-hole pair enhancing the reduction-oxidation capability of either of two semiconductors, this synergistic effect controls whether the junction will show reduced band gap due to band mismatch or individual integrity of the material for optical absorbance will be maintained. For the former, 0D/2D structure of metallic nature decorated over the semiconductor shows a huge difference in the work function, the variation in dimensionality provides variation in the charge transfer channel and availability of reaction sites. While, in the latter case, two semiconductors capable of either reduction or oxidation on vertical stacking lead to a simultaneous redox reaction enhancing the overall capability of the photo-redox mechanism. [15, 16]

Stacking of bilayer in different configurations controls the interlayer orbital interaction governing the surface activation sites and optical absorbance originating from the activation of forbidden transitions promotes the interlayer and intralayer charges for enhanced photocatalytic activity. The intercalation of alkali and alkaline earth metals within the layers of $g - C_3N_4$ strengthen the Coulomb's interaction with the layer forming the interlayer charge transfer channel due to localization of charges over the intercalated atom [17], promoting charge migration and separation hence reducing the rate of recombination of the photogenerated charge carriers. [18–21] The single-atom catalyst with its unique structural features, improved absorbance in the visible spectrum, formation of intermediate states within the forbidden region, and electronegativity difference between the N atom of $g - C_3N_4$ and Metal atom establish the reactive sites for the carbon dioxide reduction reaction (CO_2RR) and nitrogen reduction reaction (N_2RR) . The enhanced charge transfer channel further enhances the feasibility of complex reaction mechanism. Exciting results on the functionalized of $g - C_3N_4$ have been seen in this thesis. A detailed literature review is available for various functionalization over $g - C_3N_4$ towards the improvement of photocatalytic efficiency. [22–25]

2 Computational Methodology

Advancements in computational facilities have empowered condensed matter physicists in solving the many body problem through formulated Schrödinger equation by Hohenberg-Kohn in 1964. [26,27] The solution of the Kohn-Sham equation was based on the electron density rather than the traditional coordinate-based electron wavefunction. The self-consistent field cycle proposed by the Sham-Kohn in 1965 based on the variational principle changed the course of history for electronic property calculations. In this thesis, we have performed all the calculations within the density functional theory (DFT) based on a plane wave-based method and full potential linearized augmented plane waves-based solutions through Quantum Espresso (QE) [28] and Wein2K code, [29] respectively. The accurate band gap for the semiconductor material was performed using hybrid functional within QE and Trans Blahamodified Becke Johanson (TB-mBJ) potential within Wein2k software. Computation of optical properties have been performed using the epsilon.x subroutine of QE code. Maximally localized wannier functions (MLWF) is also considered for correct band structure calculation using the hybrid functional. conventional method of Nørskov et al., for OER/HER along with simultaneous water splitting using CI-NEB [30–35] have been considered for HER, OER and CO_2RR/N_2RR . HER has been investigated with H-atom and H_3O molecule adsorbed over the surface. OER was investigated with H_2O , *OH, *O, and *OOH as reaction intermediates to understand the rate-determining step, hence the overpotential. CO_2RR and N_2RR were performed with all the thermodynamically possible reaction intermediates and various reaction pathways were undertaken including the possible intermediates formed as the reaction proceeds. CO_2RR and N_2RR were investigated on the basis of the adsorption energies of the adsorbed intermediate on the single atom catalyst. [30-32, 36, 37]

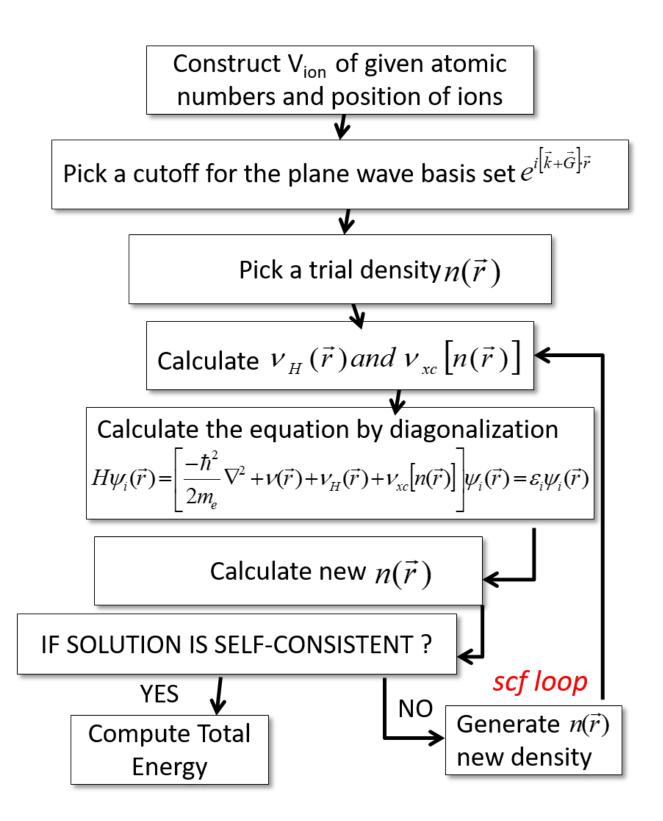


Figure 1: Self-consistent field cycle for the solution of Kohn-Sham equation.

3 Metal/Non-Metal Loaded/Doped $g - C_3N_4$ Monolayer

With the variation in the electronic properties on doping/loading of non-metal/metal atoms in the $g - C_3N_4$, the suitability of band edges for the photocatalytic reduction/oxidation could be tuned. On the backdrop of the synergistic effect seen on doping with loading, a combination of (B/Co, O/Fe, P/Ag) in $g - C_3N_4$ was investigated based on the electronic properties to screen the suitable combination for simulating the study of overall water splitting. $(Co, B) - g - C_3N_4$ was further studied to understand the role of the intermediate band on the optical absorbance and effective mass of the material to form an efficient photocatalytic material. The free energy change value confirmed the enhanced efficiency compared to the pristine monolayer overcoming the prohibitive overpotential reported for OER. [38–40]

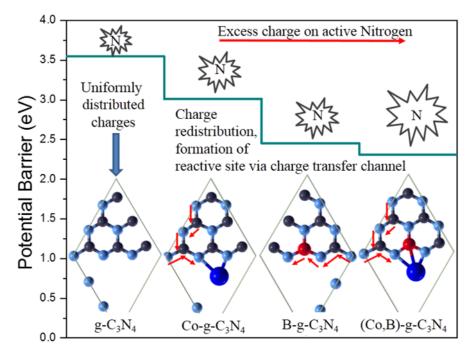


Figure 2: Relationship between overpotential and charge enhancement over the active site on B-doping, Co-loading, B-doping with Co-loading.

4 $g - C_3 N_4$ based Heterostructure

Two materials with different electronic properties in close contact form a Heterostructure, based on the type of material and their interation giving rise to the culmination of interfacial properties from both materials. This interface engineering promises multifunctional characteristics of the hybrid materials. Interface charge transfer across the junction is governed by the type of contact, the work function of materials, and their dimensionality. Based on the dimensionality and based on the type of material following sub-sections are provided with a brief description:

4.1 Metal-Semiconductor Heterostructure

Cobalt boride (Co_2B_2) based metallic materials have been considered as decoration over $g - C_3N_4$ with two different dimensionalities. The study of dimension-based metallic decoration over semiconductors and its role in the directional charge transfer for tailoring photocatalytic activity has been studied. Metallic cluster of Co_2B_2 (0D) and 2D Co_2B_2 MBene over $g-C_3N_4$ forms a Schottky barrier providing directionality to electron transfer and enhancing the reduction/oxidation potential on either substrate or decoration.

4.2 Semiconductor-Semiconductor Heterostructure

Band edge position is an essential requirement of the semiconductor materials for a redox reaction, lateral contact of two semiconductors with different band edges is prone to three varieties of alignment. Charge transfer across the interface is determined by the band bending caused due to balance between the Fermi levels of both semiconductors. Here, $Hf_2CO_2/g - C_3N_4$ is investigated to study the role of band bending in the functionality of photocatalysts. [13,14]

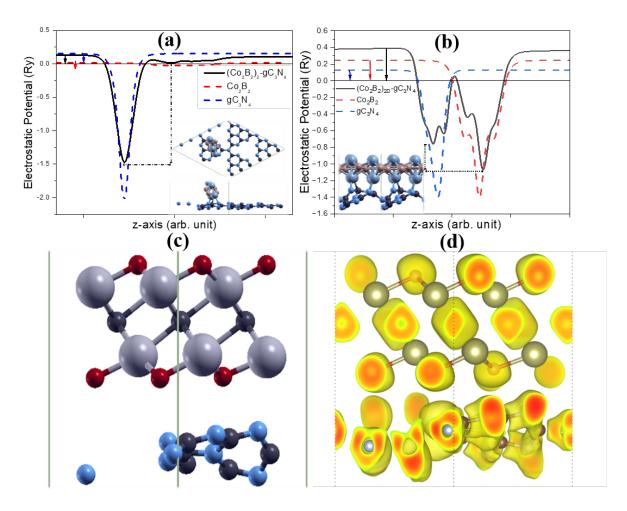


Figure 3: (a, b) Electrostatic potential profile for 0D/2D-2D Metal-Semiconductor heterojunction, (c, d) Structure and electron localization function of $Hf_2CO_2/g - C_3N_4$.

5 Modified Bilayer of $g - C_3 N_4$

Stacking/twisting of the bilayer modifies the interface property affecting the photocatalytic activity of a material. In this chapter, the 9 different stacked bilayers with AA and AB configuration are studied. The structural variation along with the electronic and optical properties modification emphasizes the role of interlayer orbital interaction over the overpotential of redox reaction. Intralayer charge redistribution separated the photoanodic and photocathodic sites. Further to eliminate the shortcoming of the bilayer structure and to build an interlayer bridge for easy charge migration between the layers, Li intercalated $g - C_3 N_4$ was investigated. Conventional and simultaneous OER/HER were performed and verified the results of free energy change and CI-NEB. Optical properties emphasize the activation of forbidden transitions within bands for enhanced light absorption capabilities along with a reduced rate of recombination. [17]

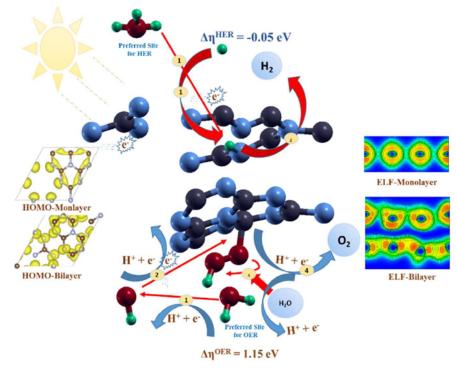


Figure 4: Interlayer and intralayer charge dynamics over suitable bilayer and activation of sites for HER/OER.

6 CO_2RR & N_2RR over Co Decorated $g - C_3N_4$

The increase of greenhouse gases, abundant nitrogen in the atmosphere, and the suitability of single-atom catalysts for adsorption for a wide range of molecules directed the path for this chapter. Among the single metal atom loaded $g - C_3N_4$ studied in Chapter 3, and receptivity of Co atom seen from its role in electrocatalyst, $Co - g - C_3N_4$ was screened for CO_2RR and N_2RR . Possible reaction intermediates formed over the surface with different adsorption sites were simulated and the reaction pathway was devised/anticipated. The most feasible reaction mechanism on the basis of adsorption energy over the suitable site confirmed the complete reduction of CO_2 and N_2 into fuel.

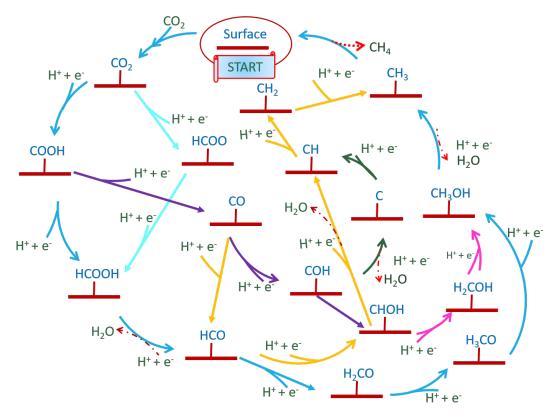


Figure 5: Possible reaction intermediates and pathways for carbon dioxide reduction to methane/methanol.

7 Conclusions and Future Prospects

This chapter comprises of a compilation of results and discussions contained in the preceding chapters on the canvas of the objectives proposed at the beginning of this thesis. The key features of the thesis will be mentioned in brief with the spotlight on the different functionalization techniques and the results obtained from the *ab-initio* calculations. With modification in the basic structure of $g - C_3N_4$, the variation in the electronic and optical properties has been explored leading to the discovery of new materials as efficient photocatalysts. In the successive parts of the chapter discussed different ways to explore the reaction mechanism over modified $g - C_3N_4$ and the future prospects. The scope of exploration from the perspective of excitonic effects will provide the conclusions of this thesis with a new perspective for understanding the inherent nature responsible for the improvement of the efficiency of photocatalysts.

Accolades and Achievements

- University, Faculty, and Department coursework have been completed successfully and passed with flying colors.
- Participated and presented poster in **VII Rajasthan Science Congress**, Mohanlal Sukhadia University, 2019. on "Enhancing photocatalytic behavior of ZnS: A Computational Perspective", won **Prof. R.S.Paroda Best Presentation Award**
- Participated and gave an oral presentation in National Research Scholars' Meet
 on Condensed Matter Physics and Materials Science (CMPMS-23), Gujarat
 University, 2023. on "Possible CO₂ reduction pathways on single atom catalyst (Cog C₃N₄)", won Best Oral Presentation Award

Publications related to thesis

- B.R. Bhagat, Alpa Dashora "Investigating dimensionality dependence of overpotential for HER/OER over $g - C_3 N_4$ based metal-semiconductor heterojunction", Under Drafting (2023).
- **B.R. Bhagat**, Alpa Dashora " $Hf_2CO_2/g C_3N_4$ semiconductor-semiconductor heterostructure for enhanced overall water splitting: An *ab-initio* study", Under Drafting (2023)
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